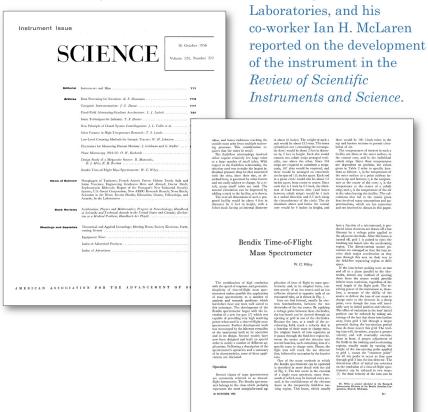
In a time not that long ago and a place not too far away, La time-of-flight mass spectrometer for commercial sale was introduced by the Cincinnati Division of Bendix, an automotive, aerospace and industrial company. This was the first successful commercial instrument that was not based on magnetic sector mass analyzer technology. Preceding the introduction of the instrument, William C. Wiley, then at the Bendix Aviation Corporation Research



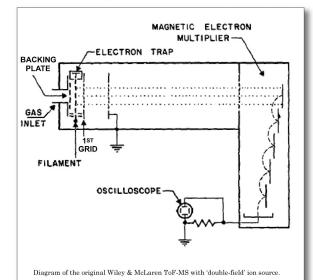
The idea for a mass spectrometer based on the difference in time that it would take ions of different mass to traverse a field-free region given the same kinetic energy was explored for over a decade before the Bendix instrument was introduced. Mass spectrometry was an important tool in the Manhattan Project and designed by Nier were used extensively, other mass analyze technologies were explored as part of the Project. William

Stephens of the University of Pennsylvania reported on a linear time-of-flight instrument at the April meeting of the American Physical Society in 1946 and followed that with a paper in Review of Scientific Instruments in 1953.

1946

Design

he problem that Wiley and McLaren saw as limiting the development of a commercial time-of-flight instrument was the need to accelerate discrete ion bunches to kilovolt energies using the state-of-the-art electronics available at that time. Instead of trying to accelerate the ion bunch in a 'single-field' source by pulsing a grid through a large potential difference, thousands of

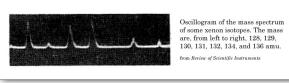


In addition, the 'double-field' source enabled

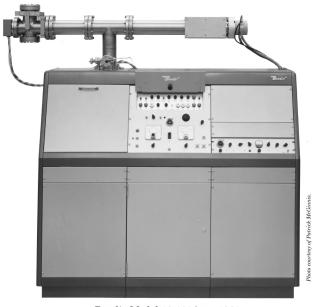
the use of an ion focusing technique that the

volts, they devised the 'double-field' source which had fixed high voltages and applied a low voltage pushout pulse on the grid at the back of the ion formation region.

The backing plate, filament, trap and 1st grid were at accelerating potential and an ion bunch was created when a pulse of electrons was accelerated across the source to the electron trap. Subsequently, a pushout pulse on the backing plate accelerated ions into the region between the two grids where they were accelerated to their final energy. Note the field-free region in this original design was at ground while the source was at accelerating voltage.



authors called 'Time-Lag Energy Focusing'; known today as delayed extraction. By varying the time between the pulse of ionizing electrons and the pushout pulse, the conditions for combined space and energy focusing could be optimized for ions over a select mass range.



Bendix Model 12-101A; ca 1961

Detection

on detection presented a serious challenge as well. In particular, ion packets arriving at the detector had peak widths on the order of 10 nsec and contained an extremely small amount of charge. An electron multiplier with 100 MHz response and amplification on the order of a million or more was required. Bendix developed their own electron multiplier using continuous dynodes immersed in an orthogonal magnetic field. This device was capable of providing both the frequency response and gain necessary to obtain mass spectra from the instrument.

Mass spectrum of methyl stearate in continuous mode operation.

primary applications of commercial mass

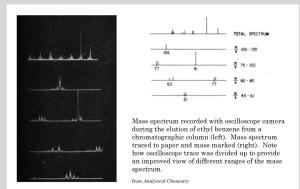
spectrometers was in the field of petrochemistry.

vacuum envelope. It had both a lens and a set of deflection plates in the field-free region to focus the ions onto the detector. The instrument operated at a fixed frequency of 10kHz and had a mass range of ~1800 u It used an electron ionization source with trap current regulation. Under these design conditions, the instrument produced hydrocarbon mass spectra very similar to those of contemporary magnetic sector instruments. This was an important goal for Bendix as one of the

The coupling of a gas chromatograph to a mass spectrometer was demonstrated in 1957 by Holmes and Merrill for a magnetic sector instrument. However, they could only observe a 16 mass unit portion of the spectrum on an oscilloscope. Gohlke also reported a GC/MS combination instrument based on the Bendix ToF-MS in 1957 followed by a more detailed

description in 1959. While mass spectra were still displayed on an oscilloscope, the mass range was as broad as the instrument could produce, typically 6000 amu with the instrument operating at a frequency of 2 kHz.

The subsequent development of the output scanner (see maintext) provided a means of oscillographic recording the complete mass spectrum of individual GC peaks in real time, on the order of several seconds per spectrum. This was a significant improvement over any other combined GC/MS instrument of the time since sector instruments took at least an order of magnitude longer to scan



Recording

nitially, the primary output of the instrument was the oscilloscope display. Permanent records L of the spectrum could be obtained with the use of an oscilloscope camera, but this was not an optimum recording medium. To remedy this problem, an output scanner, based on the concept of a 'boxcar integrator', was designed such that it would integrate the output of the electron

multiplier for a small fraction of each spectrum. By continuously shifting the integration period to later portions of subsequent spectra, a signal, comparable to that from sector instruments, could be obtained for oscillographic recording.

After a series of prototype designs, one of which was placed into service in Fred Mc Lafferty's lab at Dow Chemical in Midland Michigan, the Model 12-101 was produced and marketed. Don Damoth described this instrument in 1965 in Advances in Analytical Chemistry and

Instrumentation. It had a 1.7 m field-free region and operated at an accelerating potential of 2.8 kVdc. The ion source was at ground potential and the field-free region was maintained at accelerating potential by means of a cylindrical screen with glass standoffs inserted inside the

Schematic drawing of time-of-flight mass spectro

Demise

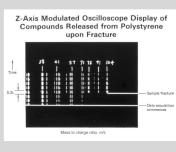
ven though the Bendix ToF-MS was well-suited for application as a GC/MS instrument and had been used to develop the technique, sales of the instrument declined in the early '70s. With the introduction of the quadrupole mass filter by Finnigan Corporation as an analytical mass spectrometer, sales of Bendix instruments began to drop off. Quads took over a larger and larger share of the 'fast scanning' instrument market after Hewlett Packard began marketing their quadrupole mass filter and eventually Bendix became part of Consolidated Vacuum Corporation, a manufacturer of vacuum systems and components. Interest in the analytical instrument part of CVC waned and eventually died out. By the mid '70s most people in the community considered mass analyzers based on the time-of-flight concept to be completely forgotten and useless; never to be heard from again!

While the Bendix instrument was not well-received in the community of sector instrument users, its unique ability to produce spectra on the millisecond time scale was attractive to a different community of scientists. An added feature that may or may not have been intentional in the instrument design was the large open ion source and the easy accessibility to the ion source housing which hung over the end of the instrument chassis. Simple vacuum flanges that could be easily modified to accommodate new and unusual experiments made the instrument particularly popular for those adventurous souls wanting to apply mass spectrometry to new areas of chemistry.

More often than not, abstracts of journal articles and conference proceedings would contain the phrase "a modified Bendix Time-of-Flight mass spectrometer was used to . . . " or "a Bendix time-of-flight mass spectrometer has been modified to . . ". Besides the development of GC/MS, the instrument was applied to a wide variety of problems:

- Investigation of thermodynamic properties of inorganic compounds by modifying the ion source to accommodate a Knudsen cell
- · Studies of photochemical oxidation Studies of ion-molecule or neutral particle-molecule interactions
- Investigation of noble gas compounds
- Chemi-ionization reactions in high-temperature hydrocarbon oxidation • Determination of chemical species prevalent in a plasma jet
- Analysis of ablator blow-off gases from space-craft heat shields
- Investigation of reactions driven by flash photolysis · Studies of ion-molecule reaction cross-sections
- · Laser ionization of organic solids
- Thermal decomposition behavior of perchlorate oxidizers studied by flash mass thermal analysis
- · Fast-flow kinetic studies of methane reactions with atomic oxygen at room temperature
- Investigation of compounds produced during polymer fracture

1979



Acknowledgement: Research supported by NIH National Center for Research Resources Grant No. P41RR095

MASS SPECTROMETRY

1948

1962

of organic solids by laser irradiation

1968

1973

analysis are published in the Federal